

### CERTIFICATE OF ANALYSIS FOR

# Gold Ore (Frogs Leg Gold Mine, Western Australia)

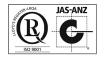
# CERTIFIED REFERENCE MATERIAL OREAS 234

Absolute Standard Deviations     Relative Standard Deviations     5% window												
Constituent	Certified		Absolute	Standard	Deviation	6	Relative	Standard D	eviations	5% w	5% window	
Value	1SD	2SD Low	2SD High	3SD Low	3SD High	1RSD	2RSD	3RSD	Low	High		
Pb Fire Assay												
Au, ppm	1.20	0.030	1.14	1.26	1.12	1.29	2.46%	4.92%	7.38%	1.14	1.26	
Aqua Regia D	igestion (sa	mple wei	ghts 10-5	0g)								
Au, ppm	1.12	0.058	1.00	1.23	0.94	1.29	5.19%	10.37%	15.56%	1.06	1.17	
Cyanide Leac	h											
Au, ppm	1.17	0.064	1.04	1.29	0.97	1.36	5.48%	10.95%	16.43%	1.11	1.22	
X-ray Photon	Assay											
Au, ppm	1.16	0.05	1.07	1.26	1.02	1.31	4.06%	8.12%	12.17%	1.11	1.22	
Aqua Regia D	igestion											
Ag, ppm	0.341	0.029	0.283	0.399	0.254	0.428	8.51%	17.01%	25.52%	0.324	0.358	
AI, wt.%	3.53	0.171	3.18	3.87	3.01	4.04	4.85%	9.69%	14.54%	3.35	3.70	
As, ppm	55	2.0	51	59	49	61	3.59%	7.18%	10.77%	52	58	
B, ppm	19.8	3.9	12.0	27.5	8.2	31.3	19.51%	39.03%	58.54%	18.8	20.7	
Ba, ppm	29.6	2.07	25.5	33.8	23.4	35.8	6.98%	13.96%	20.94%	28.2	31.1	
Be, ppm	0.27	0.021	0.22	0.31	0.20	0.33	7.90%	15.80%	23.69%	0.25	0.28	

### Table 1. Certified Values and Performance Gates for OREAS 234.

SI unit equivalents: ppm (parts per million)  $\equiv$  mg/kg  $\equiv$  µg/g  $\equiv$  0.0001 wt.%  $\equiv$  1000 ppb (parts per billion).

Note 1: intervals may appear asymmetric due to rounding.



Constituent	Certified		Absolute	Standard	Deviation	5	Relative	Standard D	eviations	5% w	indow
Constituent	Value	1SD	2SD Low	2SD High	3SD Low	3SD High	1RSD	2RSD	3RSD	Low	High
Aqua Regia D	igestion co	ntinued									
Bi, ppm	0.045	0.005	0.034	0.056	0.029	0.061	12.08%	24.17%	36.25%	0.043	0.047
Ca, wt.%	2.87	0.38	2.11	3.64	1.72	4.02	13.35%	26.69%	40.04%	2.73	3.02
Cd, ppm	0.54	0.028	0.49	0.60	0.46	0.63	5.07%	10.14%	15.20%	0.52	0.57
Ce, ppm	10.1	0.56	9.0	11.2	8.4	11.8	5.59%	11.17%	16.76%	9.6	10.6
Co, ppm	29.3	1.43	26.4	32.1	25.0	33.6	4.88%	9.76%	14.64%	27.8	30.7
Cr, ppm	21.7	1.74	18.2	25.1	16.4	26.9	8.03%	16.06%	24.09%	20.6	22.7
Cs, ppm	0.67	0.045	0.58	0.76	0.53	0.80	6.70%	13.41%	20.11%	0.63	0.70
Cu, ppm	174	7	161	187	154	194	3.85%	7.71%	11.56%	165	183
Dy, ppm	2.06	0.180	1.70	2.42	1.52	2.60	8.76%	17.52%	26.28%	1.96	2.16
Er, ppm	1.28	0.18	0.91	1.64	0.73	1.82	14.24%	28.47%	42.71%	1.21	1.34
Eu, ppm	0.42	0.06	0.31	0.53	0.25	0.59	13.53%	27.06%	40.59%	0.40	0.44
Fe, wt.%	5.39	0.194	5.00	5.77	4.80	5.97	3.61%	7.22%	10.83%	5.12	5.65
Ga, ppm	10.1	0.88	8.4	11.9	7.5	12.8	8.71%	17.41%	26.12%	9.6	10.6
Gd, ppm	1.97	0.24	1.49	2.45	1.24	2.70	12.30%	24.60%	36.91%	1.87	2.07
Ge, ppm	0.11	0.02	0.06	0.15	0.04	0.17	21.30%	42.59%	63.89%	0.10	0.11
Hf, ppm	0.51	0.09	0.34	0.69	0.25	0.77	16.83%	33.66%	50.49%	0.49	0.54
Ho, ppm	0.42	0.04	0.33	0.51	0.28	0.55	10.57%	21.14%	31.71%	0.40	0.44
In, ppm	0.036	0.004	0.027	0.045	0.023	0.049	12.26%	24.53%	36.79%	0.034	0.038
K, wt.%	0.157	0.013	0.131	0.182	0.118	0.195	8.17%	16.34%	24.51%	0.149	0.165
La, ppm	4.37	0.189	3.99	4.74	3.80	4.93	4.32%	8.63%	12.95%	4.15	4.58
Li, ppm	10.1	0.76	8.6	11.6	7.9	12.4	7.48%	14.95%	22.43%	9.6	10.6
Lu, ppm	0.16	0.02	0.12	0.20	0.10	0.22	12.09%	24.19%	36.28%	0.15	0.17
Mg, wt.%	1.63	0.063	1.50	1.75	1.44	1.82	3.87%	7.74%	11.61%	1.55	1.71
Mn, wt.%	0.074	0.004	0.065	0.082	0.060	0.087	6.06%	12.11%	18.17%	0.070	0.077
Mo, ppm	1.39	0.061	1.26	1.51	1.20	1.57	4.40%	8.80%	13.20%	1.32	1.45
Na, wt.%	0.217	0.014	0.189	0.244	0.176	0.257	6.26%	12.52%	18.78%	0.206	0.227
Nd, ppm	6.02	0.65	4.71	7.32	4.06	7.98	10.85%	21.69%	32.54%	5.72	6.32
Ni, ppm	55	4.7	46	65	41	69	8.43%	16.86%	25.30%	53	58
P, wt.%	0.039	0.002	0.035	0.043	0.033	0.045	5.19%	10.38%	15.57%	0.037	0.041
Pb, ppm	26.2	1.82	22.6	29.9	20.8	31.7	6.92%	13.84%	20.76%	24.9	27.6
Pr, ppm	1.32	0.119	1.09	1.56	0.97	1.68	8.96%	17.93%	26.89%	1.26	1.39
Pt, ppb	13.9	0.9	12.2	15.7	11.3	16.6	6.34%	12.68%	19.03%	13.2	14.6
Rb, ppm	6.39	0.620	5.15	7.63	4.53	8.25	9.70%	19.40%	29.10%	6.07	6.71
Re, ppm	0.002	0.000	0.001	0.003	0.000	0.003	27.49%	54.98%	82.47%	0.002	0.002
S, wt.%	0.370	0.026	0.319	0.421	0.293	0.447	6.95%	13.91%	20.86%	0.351	0.388
Sb, ppm	0.81	0.24	0.34	1.28	0.10	1.52	29.14%	58.27%	87.41%	0.77	0.85
Sc, ppm	6.46	0.505	5.45	7.47	4.95	7.97	7.82%	15.63%	23.45%	6.14	6.78
Sm, ppm	1.56	0.116	1.33	1.79	1.22	1.91	7.41%	14.81%	22.22%	1.48	1.64
Sn, ppm	0.62	0.08	0.45	0.79	0.37	0.87	13.63%	27.25%	40.88%	0.59	0.65
Sr, ppm	48.8	7.7	33.4	64.1	25.7	71.8	15.75%	31.49%	47.24%	46.3	51.2

#### Table 1 continued.

SI unit equivalents: ppm (parts per million)  $\equiv$  mg/kg  $\equiv$  µg/g  $\equiv$  0.0001 wt.%  $\equiv$  1000 ppb (parts per billion).

Note 1: intervals may appear asymmetric due to rounding; IND = indeterminate.



	-	-									
	Certified		Absolute	Standard	Deviation	5	Relative	Standard D	eviations	5% w	indow
Constituent	Value	1SD	2SD Low	2SD High	3SD Low	3SD High	1RSD	2RSD	3RSD	Low	High
Aqua Regia D	igestion co	ntinued									
Ta, ppm	< 0.01	IND	IND	IND	IND	IND	IND	IND	IND	IND	IND
Tb, ppm	0.32	0.026	0.27	0.37	0.25	0.40	7.97%	15.94%	23.91%	0.31	0.34
Te, ppm	0.094	0.010	0.075	0.114	0.065	0.124	10.53%	21.05%	31.58%	0.090	0.099
Th, ppm	0.83	0.09	0.66	1.01	0.57	1.10	10.58%	21.17%	31.75%	0.79	0.88
Ti, wt.%	0.325	0.049	0.226	0.423	0.177	0.472	15.18%	30.37%	45.55%	0.308	0.341
TI, ppm	0.12	0.007	0.11	0.14	0.10	0.15	5.91%	11.83%	17.74%	0.12	0.13
Tm, ppm	0.18	0.02	0.14	0.22	0.12	0.23	10.63%	21.26%	31.90%	0.17	0.19
U, ppm	0.20	0.011	0.18	0.22	0.17	0.23	5.21%	10.41%	15.62%	0.19	0.21
V, ppm	136	16	105	167	89	183	11.52%	23.04%	34.55%	129	143
W, ppm	18.1	2.8	12.5	23.8	9.7	26.6	15.47%	30.93%	46.40%	17.2	19.1
Y, ppm	11.6	1.3	9.0	14.1	7.8	15.3	10.88%	21.76%	32.64%	11.0	12.1
Yb, ppm	1.12	0.12	0.88	1.37	0.76	1.49	10.87%	21.74%	32.60%	1.07	1.18
Zn, ppm	123	5	113	132	109	137	3.78%	7.57%	11.35%	117	129
Zr, ppm	17.7	1.59	14.6	20.9	13.0	22.5	8.96%	17.92%	26.88%	16.9	18.6
4-Acid Digest	ion	I				I	<u> </u>		<u> </u>		
Ag, ppm	0.338	0.015	0.308	0.368	0.293	0.382	4.41%	8.82%	13.24%	0.321	0.355
Al, wt.%	7.05	0.175	6.70	7.41	6.53	7.58	2.48%	4.97%	7.45%	6.70	7.41
As, ppm	55	3.6	48	63	45	66	6.54%	13.07%	19.61%	53	58
Ba, ppm	129	4	122	137	118	140	2.84%	5.68%	8.52%	123	136
Be, ppm	0.42	0.029	0.36	0.48	0.34	0.51	6.96%	13.91%	20.87%	0.40	0.44
Bi, ppm	0.047	0.009	0.029	0.065	0.020	0.074	19.00%	38.01%	57.01%	0.045	0.049
Ca, wt.%	7.08	0.285	6.51	7.65	6.23	7.94	4.03%	8.06%	12.09%	6.73	7.44
Cd, ppm	0.57	0.030	0.51	0.63	0.48	0.66	5.24%	10.48%	15.72%	0.54	0.59
Ce, ppm	12.9	0.72	11.5	14.3	10.7	15.0	5.56%	11.12%	16.68%	12.2	13.5
Co, ppm	41.2	2.17	36.9	45.6	34.7	47.8	5.27%	10.55%	15.82%	39.2	43.3
Cr, ppm	110	12	87	133	75	145	10.53%	21.05%	31.58%	104	115
Cs, ppm	0.92	0.040	0.84	1.00	0.80	1.04	4.33%	8.66%	12.98%	0.88	0.97
Cu, ppm	175	6	164	187	158	192	3.28%	6.57%	9.85%	166	184
Dy, ppm	3.60	0.191	3.22	3.98	3.03	4.18	5.30%	10.60%	15.90%	3.42	3.78
Er, ppm	2.17	0.087	1.99	2.34	1.91	2.43	4.03%	8.05%	12.08%	2.06	2.28
Eu, ppm	0.90	0.047	0.81	1.00	0.76	1.05	5.24%	10.49%	15.73%	0.86	0.95
Fe, wt.%	7.53	0.225	7.08	7.98	6.85	8.20	2.99%	5.98%	8.97%	7.15	7.90
Ga, ppm	15.5	0.65	14.2	16.8	13.6	17.5	4.18%	8.35%	12.53%	14.8	16.3
Gd, ppm	3.18	0.233	2.72	3.65	2.48	3.88	7.32%	14.64%	21.96%	3.02	3.34
Hf, ppm	1.64	0.103	1.44	1.85	1.34	1.95	6.25%	12.50%	18.75%	1.56	1.73
Ho, ppm	0.76	0.052	0.66	0.86	0.61	0.92	6.77%	13.55%	20.32%	0.72	0.80
In, ppm	0.074	0.009	0.056	0.092	0.047	0.100	12.04%	24.09%	36.13%	0.070	0.078
K, wt.%	0.460	0.018	0.424	0.496	0.406	0.514	3.95%	7.89%	11.84%	0.437	0.483
La, ppm	5.51	0.313	4.88	6.13	4.57	6.45	5.68%	11.36%	17.04%	5.23	5.78
Li, ppm	11.0	0.81	9.4	12.7	8.6	13.5	7.36%	14.73%	22.09%	10.5	11.6
ы, ррш	11.0	0.01	3.4	12.1	0.0	10.0	1.0070	1.10/0	22.0370	10.0	11.0

#### Table 1 continued.

SI unit equivalents: ppm (parts per million)  $\equiv$  mg/kg  $\equiv$  µg/g  $\equiv$  0.0001 wt.%  $\equiv$  1000 ppb (parts per billion).

Note 1: intervals may appear asymmetric due to rounding.



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Constituent	Certified		Absolute	Standard	Deviation	6	Relative	Standard D	eviations	5% w	indow
Constituent	Value	1SD	2SD Low	2SD High	3SD Low	3SD High	1RSD	2RSD	3RSD	Low	High
4-Acid Digest	ion continue	∋d									
Lu, ppm	0.31	0.019	0.28	0.35	0.26	0.37	5.95%	11.90%	17.85%	0.30	0.33
Mg, wt.%	3.59	0.125	3.34	3.84	3.21	3.97	3.49%	6.99%	10.48%	3.41	3.77
Mn, wt.%	0.130	0.004	0.122	0.138	0.118	0.142	3.09%	6.18%	9.27%	0.123	0.136
Mo, ppm	1.49	0.108	1.28	1.71	1.17	1.82	7.21%	14.42%	21.62%	1.42	1.57
Na, wt.%	1.64	0.069	1.51	1.78	1.44	1.85	4.22%	8.44%	12.65%	1.56	1.73
Nb, ppm	3.36	0.226	2.91	3.81	2.68	4.04	6.73%	13.46%	20.19%	3.19	3.53
Nd, ppm	8.32	0.370	7.58	9.06	7.21	9.43	4.45%	8.90%	13.35%	7.90	8.73
Ni, ppm	81	3.9	74	89	70	93	4.77%	9.55%	14.32%	77	85
P, wt.%	0.041	0.002	0.037	0.045	0.036	0.046	4.40%	8.80%	13.20%	0.039	0.043
Pb, ppm	26.4	1.36	23.7	29.1	22.3	30.5	5.16%	10.32%	15.48%	25.1	27.7
Pr, ppm	1.75	0.088	1.57	1.93	1.49	2.01	5.02%	10.05%	15.07%	1.66	1.84
Rb, ppm	14.0	0.75	12.5	15.4	11.7	16.2	5.34%	10.68%	16.02%	13.3	14.6
S, wt.%	0.376	0.028	0.320	0.431	0.293	0.459	7.37%	14.75%	22.12%	0.357	0.395
Sb, ppm	1.50	0.100	1.30	1.70	1.20	1.80	6.69%	13.39%	20.08%	1.42	1.57
Sc, ppm	38.7	1.70	35.3	42.1	33.6	43.8	4.40%	8.80%	13.20%	36.8	40.7
Sm, ppm	2.43	0.097	2.24	2.62	2.14	2.72	3.99%	7.98%	11.96%	2.31	2.55
Sn, ppm	0.98	0.096	0.78	1.17	0.69	1.26	9.85%	19.69%	29.54%	0.93	1.02
Sr, ppm	211	10	192	230	182	240	4.59%	9.18%	13.78%	200	221
Ta, ppm	0.23	0.02	0.18	0.28	0.15	0.30	10.94%	21.88%	32.81%	0.22	0.24
Tb, ppm	0.55	0.040	0.47	0.63	0.43	0.67	7.33%	14.66%	21.99%	0.52	0.57
Te, ppm	0.096	0.015	0.065	0.127	0.049	0.142	16.14%	32.28%	48.42%	0.091	0.101
Th, ppm	1.03	0.073	0.89	1.18	0.81	1.25	7.07%	14.14%	21.20%	0.98	1.08
Ti, wt.%	0.587	0.022	0.543	0.631	0.521	0.653	3.75%	7.51%	11.26%	0.558	0.616
TI, ppm	0.23	0.013	0.21	0.26	0.20	0.27	5.46%	10.91%	16.37%	0.22	0.25
Tm, ppm	0.31	0.015	0.28	0.34	0.27	0.36	4.91%	9.82%	14.74%	0.30	0.33
U, ppm	0.29	0.020	0.25	0.33	0.23	0.35	7.06%	14.12%	21.18%	0.28	0.30
V, ppm	263	9	244	281	235	290	3.45%	6.91%	10.36%	249	276
W, ppm	26.2	1.62	23.0	29.5	21.4	31.1	6.19%	12.38%	18.57%	24.9	27.5
Y, ppm	19.5	0.93	17.6	21.3	16.7	22.2	4.76%	9.52%	14.28%	18.5	20.4
Yb, ppm	2.14	0.112	1.92	2.37	1.81	2.48	5.25%	10.49%	15.74%	2.04	2.25
Zn, ppm	135	7	120	150	113	157	5.53%	11.06%	16.60%	128	142
Zr, ppm	55	4.5	46	63	41	68	8.20%	16.41%	24.61%	52	57

### Table 1 continued.

SI unit equivalents: ppm (parts per million)  $\equiv$  mg/kg  $\equiv$  µg/g  $\equiv$  0.0001 wt.%  $\equiv$  1000 ppb (parts per billion).

Note 1: intervals may appear asymmetric due to rounding.



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Constituent	Unit	Value	Constituent	Unit	Value	Constituent	Unit	Value
Pb Fire Assa	ay		<u> </u>		I			
Pd	ppb	10.0	Pt	ppb	15.0			
Aqua Regia	Digestio	n						
Hg	ppm	0.037	Pd	ppb	11.8	Si	wt.%	0.110
Nb	ppm	0.19	Se	ppm	0.72			
Borate Fusio	on XRF					•		
Al <sub>2</sub> O <sub>3</sub>	wt.%	13.43	MgO	wt.%	6.15	SiO <sub>2</sub>	wt.%	51.87
CaO	wt.%	10.18	MnO	wt.%	0.170	SO <sub>3</sub>	wt.%	0.904
Fe <sub>2</sub> O <sub>3</sub>	wt.%	10.97	Na <sub>2</sub> O	wt.%	2.16	TiO <sub>2</sub>	wt.%	1.01
K <sub>2</sub> O	wt.%	0.547	P <sub>2</sub> O <sub>5</sub>	wt.%	0.096			
4-Acid Diges	stion							
В	ppm	1157	Hg	ppm	< 2	Se	ppm	0.94
Ge	ppm	0.17	Re	ppm	0.002			
Thermograv	imetry					•		
LOI <sup>1000</sup>	wt.%	3.35						
Infrared Con	nbustion					•		
С	wt.%	0.125	S	wt.%	0.340			
Laser Ablati	on ICP-N	IS				•		
Ag	ppm	0.400	Hf	ppm	2.05	Sm	ppm	2.52
As	ppm	55	Но	ppm	0.81	Sn	ppm	1.00
Ва	ppm	135	In	ppm	0.050	Sr	ppm	209
Be	ppm	0.60	La	ppm	5.73	Та	ppm	0.22
Bi	ppm	0.050	Lu	ppm	0.34	Tb	ppm	0.59
Cd	ppm	0.65	Mn	wt.%	0.135	Те	ppm	< 0.2
Ce	ppm	12.7	Мо	ppm	1.40	Th	ppm	1.07
Со	ppm	43.6	Nb	ppm	3.54	Ti	wt.%	0.605
Cr	ppm	137	Nd	ppm	8.92	TI	ppm	< 0.2
Cs	ppm	0.89	Ni	ppm	91	Tm	ppm	0.34
Cu	ppm	191	Pb	ppm	28.0	U	ppm	0.32
Dy	ppm	3.64	Pr	ppm	1.84	V	ppm	282
Er	ppm	2.36	Rb	ppm	14.0	W	ppm	27.0
Eu	ppm	0.92	Re	ppm	< 0.01	Y	ppm	20.5
Ga	ppm	15.5	Sb	ppm	1.65	Yb	ppm	2.31
Gd	ppm	3.26	Sc	ppm	39.3	Zn	ppm	140
Ge	ppm	1.48	Se	ppm	< 5	Zr	ppm	65

### Table 2. Indicative Values for OREAS 234.

SI unit equivalents: ppm (parts per million)  $\equiv$  mg/kg  $\equiv$  µg/g  $\equiv$  0.0001 wt.%  $\equiv$  1000 ppb (parts per billion).

Note: the number of significant figures reported is not a reflection of the level of certainty of stated values. They are instead an artefact of ORE's in-house CRM-specific LIMS.

## INTRODUCTION

OREAS reference materials are intended to provide a low-cost method of evaluating and improving the quality of analysis of geological samples. To the geologist they provide a means of implementing quality control in analytical data sets generated in exploration from the grass roots level through to prospect evaluation, and in grade control at mining operations. To the analyst they provide an effective means of calibrating analytical equipment, assessing new techniques and routinely monitoring in-house procedures.



OREAS reference materials enable users to successfully achieve process control of these tasks because the observed variance from repeated analysis has its origin almost exclusively in the analytical process rather than the reference material itself. In evaluating laboratory performance with this CRM, the section headed 'Instructions for correct use' should be read carefully.

Table 1 provides performance gate intervals for the certified values based on their pooled 1SD's. Table 2 shows indicative values including major and trace element characterisation by Bureau Veritas in Perth, Western Australia which includes:

- Major oxides by lithium borate fusion with X-ray fluorescence;
- LOI at 1000°C by thermogravimetric analyser;
- Total Carbon and Sulphur by Infrared combustion furnace;
- Trace element characterisation by laser ablation with ICP-MS finish.

Table 3 provides some indicative physical properties and Table 4 presents the 95% confidence and tolerance limits for all certified values. Gold homogeneity (via INAA) is shown in Table 5 and is also demonstrated by a nested ANOVA program using fire assay (see '**nested ANOVA**' section).

Tabulated results of all elements (including Au INAA analyses) together with uncorrected means, medians, standard deviations, relative standard deviations and percent deviation of lab means from the corrected mean of means (PDM<sup>3</sup>) are presented in the detailed certification data for this CRM (**OREAS 234-DataPack.1.0.210511\_112628.xlsx**).

Results are also presented in scatter plots for gold by fire assay, aqua regia digestion and cyanide leach (Figures 1 to 3, respectively) together with  $\pm 3$ SD (magenta) and  $\pm 5\%$  (yellow) control lines and certified value (green line). Accepted individual results are coloured blue and individual and dataset outliers are identified in red and violet, respectively.

## SOURCE MATERIAL

Certified Reference Material (CRM) OREAS 234 was prepared from a blend of goldbearing ore and barren greenstone. The ore was sourced from the Frogs Leg Gold Mine located 19km west of Kalgoorlie in Western Australia. The Cambrian greenstone was sourced from a quarry 145km north of Melbourne, Australia.

# PERFORMANCE GATES

Table 1 above shows intervals calculated for two and three standard deviations. As a guide these intervals may be regarded as warning or rejection for multiple 2SD outliers, or rejection for individual 3SD outliers in QC monitoring, although their precise application should be at the discretion of the QC manager concerned (also see 'Intended Use' section below). Westgard Rules extend the basics of single-rule QC monitoring using multi-rules (for more information visit www.westgard.com/mltirule.htm). A second method utilises a 5% window calculated directly from the certified value.



Standard deviation is also shown in relative percent for one, two and three relative standard deviations (1RSD, 2RSD and 3RSD) to facilitate an appreciation of the magnitude of these numbers and a comparison with the 5% window. Caution should be exercised when concentration levels approach lower limits of detection of the analytical methods employed as performance gates calculated from standard deviations tend to be excessively wide whereas those determined by the 5% method are too narrow. One approach used at commercial laboratories is to set the acceptance criteria at twice the detection level (DL)  $\pm$  10%.

*i.e.* Certified Value ± 10% ± 2DL (adapted from Govett, 1983).

# COMMINUTION AND HOMOGENISATION PROCEDURES

The material constituting OREAS 234 was prepared in the following manner:

- Drying the gold ore and greenstone to constant mass at 105°C;
- Crushing and multi stage milling of gold ore to 100% minus 30 microns;
- Crushing and multi stage milling of greenstone to >98% minus 75 microns;
- Final homogenisation;
- Packaging in 60g units sealed in laminated foil pouches and 1kg units in plastic jars.

# PHYSICAL PROPERTIES

OREAS 234 was tested at ORE Research & Exploration Pty Ltd's onsite facility for various physical properties. Table 3 presents these findings that should be used for informational purposes only.

Bulk Density (g/L)	Moisture%	Munsell Notation <sup>‡</sup>	Munsell Color <sup>‡</sup>						
928.1	0.7	5GY 6/1	Greenish Gray						

### Table 3. Physical properties of OREAS 234.

<sup>‡</sup>The Munsell Rock Color Chart helps geologists and archeologists communicate with colour more effectively by cross-referencing ISCC-NBS colour names with unique Munsell alpha-numeric colour notations for rock colour samples.

# ANALYTICAL PROGRAM

Thirty-one commercial analytical laboratories participated in the program to certify the elements reported in Table 1. The following methods were employed:

- Gold by fire assay (15-40g charge weight) with AAS (19 laboratories), ICP-OES (11 laboratories) or ICP-MS finish (1 laboratory);
- Gold by aqua regia digestion (15-50g sample weight) with ICP-MS (14 laboratories), AAS (7 laboratories) or ICP-OES finish (1 laboratory);
- Gold by cyanide leach; a variety of cyanide leach methods were undertaken by the participating laboratories including the use of LeachWELL tablets, alkaline added sodium cyanide solution as well as sodium cyanide liquor with LeachWELL powder. The sample weights included: 20g (1 laboratory by AAS finish), 30g (7 laboratories by AAS finish), 50g (3 laboratories by ICP finish and 1 laboratory by AAS finish),



60g (1 laboratory by ICP finish) and 200g (6 laboratories by AAS and 1 laboratory by ICP finish).

- Gold by x-ray photon assay on ~350g sample weights (4 Chrysos PhotonAssay units at 3 installations with two rounds of data reported from 3 of the units);
- Full ICP-OES and ICP-MS elemental suites by aqua regia digestion (up to 25 laboratories depending on the element);
- Full ICP-OES and ICP-MS elemental suites by 4-acid (HNO<sub>3</sub>-HF-HCIO<sub>4</sub>-HCI) digestion (up to 25 laboratories depending on the element).

To confirm homogeneity, gold by instrumental neutron activation analysis (INAA) was undertaken on 20 x 85mg subsamples by the Australian Nuclear Science and Technology Organisation (ANSTO) located in Lucas Heights, NSW, Australia (see Table 5 in the 'Homogeneity Evaluation' section below).

For the round robin program twenty 3kg test units were taken at predetermined intervals during the bagging stage, immediately following homogenisation and are considered representative of the entire prepared batch. Six pulp samples were submitted to each laboratory for analysis (the weight provided depended on whether the laboratory was anticipated to undertake assays by gold cyanide leach). The samples received by each laboratory were obtained by taking two samples from each of three separate 3kg test units. This format enabled a nested ANOVA treatment of the results to evaluate homogeneity, i.e., to ascertain whether between-unit variance is greater than within-unit variance.

# STATISTICAL ANALYSIS

**Standard Deviation** intervals (see Table 1) provide an indication of a level of performance that might reasonably be expected from a laboratory being monitored by this CRM in a QA/QC program. They take into account errors attributable to measurement uncertainty and CRM variability. For an effective CRM the contribution of the latter should be negligible in comparison to measurement errors. The Standard Deviation values include all sources of measurement uncertainty: between-lab variance, within-run variance (precision errors) and CRM variability.

In the application of SD's in monitoring performance it is important to note that not all laboratories function at the same level of proficiency and that different methods in use at a particular laboratory have differing levels of precision. Each laboratory has its own inherent SD (for a specific concentration level and analyte-method pair) based on the analytical process and this SD is not directly related to the round robin program (see 'Instructions for correct use' section for more detail).

The SD for each analyte's certified value is calculated from the same filtered data set used to determine the certified value, i.e., after removal of all individual, lab dataset (batch) and 3SD outliers (single iteration). These outliers can only be removed after the absolute homogeneity of the CRM has been independently established, i.e., the outliers must be confidently deemed to be analytical rather than arising from inhomogeneity of the CRM. *The standard deviation is then calculated for each analyte from the pooled accepted analyses generated from the certification program.* 

**Certified Values, Standard Deviations, Confidence Limits and Tolerance Limits** (Table 4) have been determined for each analyte following removal of individual, laboratory dataset (batch) and 3SD outliers (single iteration).



For individual outliers within a laboratory batch the z-score test is used in combination with a second method that determines the per cent deviation of the individual value from the batch median. Outliers in general are selected on the basis of z-scores > 2.5 and with per cent deviations (i) > 3 and (ii) more than three times the average absolute per cent deviation for the batch. In certain instances, statistician's prerogative has been employed in discriminating outliers.

Each laboratory data set mean is tested for outlying status based on z-score discrimination and rejected if > 2.5. After individual and laboratory data set (batch) outliers have been eliminated a non-iterative 3 standard deviation filter is applied, with those values lying outside this window also relegated to outlying status.

**Certified Values** are the means of accepted laboratory means after outlier filtering. The INAA data (see Table 5) is omitted from determination of the certified value for Au and is used solely for the calculation of Tolerance Limits and homogeneity evaluation of OREAS 234 (see 'Homogeneity Evaluation' section below).

**95% Confidence Limits** are inversely proportional to the number of participating laboratories and inter-laboratory agreement. It is a measure of the reliability of the certified value. A 95% confidence interval indicates a 95% probability that the true value of the analyte under consideration lies between the upper and lower limits. **95% Confidence Limits should not be used as control limits for laboratory performance.** 

**Indicative (uncertified) values** (Table 2) are present where the number of laboratories reporting a particular analyte is insufficient (< 5) to support certification or where interlaboratory consensus is poor.

### Homogeneity Evaluation

For analytes other than gold, the tolerance limits (ISO 16269:2014) shown in Table 4 were determined using an analysis of precision errors method and are considered a conservative estimate of true homogeneity. The meaning of tolerance limits may be illustrated for copper by 4-acid digestion, where 99% of the time  $(1-\alpha=0.99)$  at least 95% of subsamples (p=0.95) will have concentrations lying between 171 and 179 ppm. Put more precisely, this means that if the same number of subsamples were taken and analysed in the same manner repeatedly, 99% of the tolerance intervals so constructed would cover at least 95% of the total population, and 1% of the tolerance intervals would cover less than 95% of the total population (ISO Guide 35).

# Please note that tolerance limits pertain to the homogeneity of the CRM only and should not be used as control limits for laboratory performance.



• • • •	Certified	95% Confid	dence Limits	95% Tolera	1.20*     1.21*       1.11*     1.12*       1.16*     1.17*       1.16*     1.16*		
Constituent	Value	Low	High	Value	Low		
Pb Fire Assay		L	1				
Au, Gold (ppm)	1.20	1.19	1.21	1.20*	1.21*		
Aqua Regia Digestion (s	ample weights 1	0-50g)					
Au, Gold (ppm)	1.12	1.09	1.14	1.11*	1.12*		
Cyanide Leach							
Au, Gold (ppm)	1.17	1.14	1.20	1.16*	1.17*		
X-ray Photon Assay							
Au, Gold (ppm)	1.16	1.11	1.21	1.16*	1.16*		
Aqua Regia Digestion			1				
Ag, Silver (ppm)	0.341	0.326	0.356	0.324	0.358		
Al, Aluminium (wt.%)	3.53	3.45	3.61	3.46	3.60		
As, Arsenic (ppm)	55	54	56	54	56		
B, Boron (ppm)	19.8	16.8	22.8	IND	IND		
Ba, Barium (ppm)	29.6	28.6	30.7	28.7	30.5		
Be, Beryllium (ppm)	0.27	0.26	0.27	0.23	0.30		
Bi, Bismuth (ppm)	0.045	0.042	0.048	IND	IND		
Ca, Calcium (wt.%)	2.87	2.69	3.06	2.81	2.94		
Cd, Cadmium (ppm)	0.54	0.53	0.56	0.52	0.57		
Ce, Cerium (ppm)	10.1	9.8	10.4	9.9	10.3		
Co, Cobalt (ppm)	29.3	28.7	29.8	28.6	30.0		
Cr, Chromium (ppm)	21.7	21.0	22.4	21.0	22.3		
Cs, Caesium (ppm)	0.67	0.64	0.69	0.65	0.69		
Cu, Copper (ppm)	174	171	177	170	178		
Dy, Dysprosium (ppm)	2.06	1.89	2.23	1.99	2.13		
Er, Erbium (ppm)	1.28	1.12	1.43	1.20	1.35		
Eu, Europium (ppm)	0.42	0.36	0.48	0.40	0.44		
Fe, Iron (wt.%)	5.39	5.29	5.48	5.29	5.48		
Ga, Gallium (ppm)	10.1	9.7	10.5	9.8	10.4		
Gd, Gadolinium (ppm)	1.97	1.76	2.18	1.89	2.05		
Ge, Germanium (ppm)	0.11	0.08	0.13	IND	IND		
Hf, Hafnium (ppm)	0.51	0.47	0.56	0.49	0.54		
Ho, Holmium (ppm)	0.42	0.38	0.46	0.40	0.43		
In, Indium (ppm)	0.036	0.034	0.038	0.030	0.042		
K, Potassium (wt.%)	0.157	0.151	0.162	0.154	0.160		
La, Lanthanum (ppm)	4.37	4.28	4.45	4.27	4.46		
Li, Lithium (ppm)	10.1	9.7	10.5	9.8	10.5		
Lu, Lutetium (ppm)	0.16	0.14	0.18	IND	IND		

Table 4. 95% Confidence & Tolerance Limits for OREAS 234.

SI unit equivalents: ppm (parts per million)  $\equiv$  mg/kg  $\equiv$  µg/g  $\equiv$  0.0001 wt.%  $\equiv$  1000 ppb (parts per billion).

\*Gold Tolerance Limits for typical 30g fire assay, 25g aqua regia digestion, 200g cyanide leach and 350g x-ray PhotonAssay methods are determined from 20 x 85mg INAA results and the Sampling Constant (Ingamells & Switzer, 1973).

Note: intervals may appear asymmetric due to rounding.



O and the sect	Certified	95% Confid	ence Limits	95% Tolerance Lim		
Constituent	Value	Low	High	Value	Low	
Aqua Regia Digestion con	tinued				L	
Mg, Magnesium (wt.%)	1.63	1.60	1.66	1.60	1.66	
Mn, Manganese (wt.%)	0.074	0.072	0.075	0.072	0.075	
Mo, Molybdenum (ppm)	1.39	1.36	1.41	1.32	1.45	
Na, Sodium (wt.%)	0.217	0.211	0.222	0.209	0.224	
Nd, Neodymium (ppm)	6.02	5.45	6.58	5.81	6.23	
Ni, Nickel (ppm)	55	53	57	54	57	
P, Phosphorus (wt.%)	0.039	0.038	0.040	0.038	0.040	
Pb, Lead (ppm)	26.2	25.5	27.0	25.5	27.0	
Pr, Praseodymium (ppm)	1.32	1.19	1.46	1.28	1.37	
Pt, Platinum (ppb)	13.9	13.3	14.6	IND	IND	
Rb, Rubidium (ppm)	6.39	6.10	6.69	6.23	6.56	
Re, Rhenium (ppm)	0.002	0.002	0.002	IND	IND	
S, Sulphur (wt.%)	0.370	0.358	0.382	0.360	0.380	
Sb, Antimony (ppm)	0.81	0.69	0.93	0.78	0.85	
Sc, Scandium (ppm)	6.46	6.19	6.73	6.24	6.68	
Sm, Samarium (ppm)	1.56	1.45	1.67	1.51	1.61	
Sn, Tin (ppm)	0.62	0.57	0.67	0.60	0.64	
Sr, Strontium (ppm)	48.8	45.4	52.2	47.2	50.4	
Ta, Tantalum (ppm)	< 0.01	IND	IND	IND	IND	
Tb, Terbium (ppm)	0.32	0.30	0.34	0.31	0.33	
Te, Tellurium (ppm)	0.094	0.090	0.099	IND	IND	
Th, Thorium (ppm)	0.83	0.76	0.91	0.82	0.85	
Ti, Titanium (wt.%)	0.325	0.301	0.348	0.315	0.334	
TI, Thallium (ppm)	0.12	0.12	0.13	IND	IND	
Tm, Thulium (ppm)	0.18	0.16	0.20	0.17	0.19	
U, Uranium (ppm)	0.20	0.20	0.21	0.19	0.22	
V, Vanadium (ppm)	136	128	144	133	138	
W, Tungsten (ppm)	18.1	16.9	19.4	17.7	18.6	
Y, Yttrium (ppm)	11.6	10.9	12.2	11.2	11.9	
Yb, Ytterbium (ppm)	1.12	1.00	1.24	1.06	1.18	
Zn, Zinc (ppm)	123	121	125	120	125	
Zr, Zirconium (ppm)	17.7	16.9	18.6	17.2	18.3	
4-Acid Digestion						
Ag, Silver (ppm)	0.338	0.333	0.343	0.316	0.359	
Al, Aluminium (wt.%)	7.05	6.99	7.12	6.94	7.17	
As, Arsenic (ppm)	55	54	57	53	58	
Ba, Barium (ppm)	129	128	131	125	133	
Be, Beryllium (ppm)	0.42	0.41	0.44	0.37	0.47	

SI unit equivalents: ppm (parts per million)  $\equiv$  mg/kg  $\equiv$  µg/g  $\equiv$  0.0001 wt.%  $\equiv$  1000 ppb (parts per billion).



Operatives t	Certified	95% Confid	lence Limits	95% Tolera	ance Limits
Constituent	Value	Low	High	Value	Low
4-Acid Digestion continue	d			1	
Bi, Bismuth (ppm)	0.047	0.043	0.051	IND	IND
Ca, Calcium (wt.%)	7.08	6.97	7.20	6.98	7.18
Cd, Cadmium (ppm)	0.57	0.55	0.58	0.54	0.59
Ce, Cerium (ppm)	12.9	12.6	13.2	12.6	13.2
Co, Cobalt (ppm)	41.2	40.3	42.1	40.3	42.1
Cr, Chromium (ppm)	110	105	115	106	114
Cs, Caesium (ppm)	0.92	0.90	0.94	0.87	0.97
Cu, Copper (ppm)	175	173	177	171	179
Dy, Dysprosium (ppm)	3.60	3.49	3.71	3.48	3.72
Er, Erbium (ppm)	2.17	2.12	2.22	2.10	2.24
Eu, Europium (ppm)	0.90	0.87	0.94	0.88	0.93
Fe, Iron (wt.%)	7.53	7.44	7.62	7.40	7.66
Ga, Gallium (ppm)	15.5	15.3	15.8	15.1	16.0
Gd, Gadolinium (ppm)	3.18	3.05	3.32	3.06	3.31
Hf, Hafnium (ppm)	1.64	1.60	1.69	1.57	1.72
Ho, Holmium (ppm)	0.76	0.72	0.80	0.73	0.79
In, Indium (ppm)	0.074	0.071	0.076	0.067	0.080
K, Potassium (wt.%)	0.460	0.452	0.468	0.450	0.470
La, Lanthanum (ppm)	5.51	5.37	5.65	5.37	5.65
Li, Lithium (ppm)	11.0	10.7	11.4	10.6	11.5
Lu, Lutetium (ppm)	0.31	0.30	0.32	0.30	0.33
Mg, Magnesium (wt.%)	3.59	3.54	3.64	3.53	3.65
Mn, Manganese (wt.%)	0.130	0.128	0.131	0.128	0.132
Mo, Molybdenum (ppm)	1.49	1.45	1.54	1.38	1.60
Na, Sodium (wt.%)	1.64	1.61	1.67	1.62	1.67
Nb, Niobium (ppm)	3.36	3.25	3.46	3.23	3.48
Nd, Neodymium (ppm)	8.32	8.09	8.54	8.10	8.53
Ni, Nickel (ppm)	81	80	83	80	83
P, Phosphorus (wt.%)	0.041	0.040	0.042	0.040	0.042
Pb, Lead (ppm)	26.4	25.9	27.0	25.7	27.1
Pr, Praseodymium (ppm)	1.75	1.70	1.80	1.69	1.81
Rb, Rubidium (ppm)	14.0	13.7	14.2	13.6	14.3
S, Sulphur (wt.%)	0.376	0.364	0.388	0.366	0.385
Sb, Antimony (ppm)	1.50	1.45	1.54	1.44	1.56
Sc, Scandium (ppm)	38.7	38.1	39.4	37.5	40.0
Sm, Samarium (ppm)	2.43	2.38	2.49	2.35	2.51
Sn, Tin (ppm)	0.98	0.93	1.02	IND	IND
Sr, Strontium (ppm)	211	207	215	206	216

SI unit equivalents: ppm (parts per million)  $\equiv$  mg/kg  $\equiv$  µg/g  $\equiv$  0.0001 wt.%  $\equiv$  1000 ppb (parts per billion). Note: intervals may appear asymmetric due to rounding.



l able 4 continued.						
Constituent	Certified	95% Confic	95% Confidence Limits		95% Tolerance Limits	
	Value	Low	High	Value	Low	
4-Acid Digestion contin	ued					
Ta, Tantalum (ppm)	0.23	0.21	0.24	0.21	0.25	
Tb, Terbium (ppm)	0.55	0.52	0.58	0.53	0.57	
Te, Tellurium (ppm)	0.096	0.088	0.104	IND	IND	
Th, Thorium (ppm)	1.03	1.00	1.06	1.00	1.06	
Ti, Titanium (wt.%)	0.587	0.578	0.596	0.570	0.604	
Tl, Thallium (ppm)	0.23	0.23	0.24	0.22	0.25	
Tm, Thulium (ppm)	0.31	0.30	0.32	0.30	0.33	
U, Uranium (ppm)	0.29	0.28	0.30	0.28	0.30	
V, Vanadium (ppm)	263	259	266	254	271	
W, Tungsten (ppm)	26.2	25.5	27.0	25.7	26.8	
Y, Yttrium (ppm)	19.5	19.1	19.8	19.0	19.9	
Yb, Ytterbium (ppm)	2.14	2.08	2.20	2.06	2.22	
Zn, Zinc (ppm)	135	132	138	132	138	
Zr, Zirconium (ppm)	55	53	56	52	57	

Table 4 continued.

SI unit equivalents: ppm (parts per million)  $\equiv$  mg/kg  $\equiv$  µg/g  $\equiv$  0.0001 wt.%  $\equiv$  1000 ppb (parts per billion). Note: intervals may appear asymmetric due to rounding.

Table 5 below shows the gold INAA data determined on 20 x 85mg subsamples of OREAS 234. An equivalent scaled version of the results is also provided to demonstrate the level of repeatability that would be achieved if 30g fire assay determinations were undertaken without the normal measurement error associated with this methodology. The homogeneity of gold has been determined by INAA using the reduced analytical subsample method which utilises the known relationship between standard deviation and analytical subsample weight (Ingamells and Switzer, 1973). In this approach the sample aliquot is substantially reduced to a point where most of the variability in replicate assays should be due to inhomogeneity of the reference material (i.e., sampling error) and measurement error becomes negligible. In this instance a subsample weight of 85 milligrams was employed and the 1RSD of 0.135% was calculated for a 30g fire assay sample (2.52% at 85mg weights) and confirms the high level of gold homogeneity in OREAS 234.



Replicate	Au	Au
No	85mg actual	30g equivalent*
1	1.239	1.211
2	1.198	1.209
3	1.224	1.210
4	1.166	1.207
5	1.216	1.210
6	1.238	1.211
7	1.207	1.210
8	1.222	1.210
9	1.226	1.211
10	1.155	1.207
11	1.198	1.209
12	1.235	1.211
13	1.216	1.210
14	1.160	1.207
15	1.252	1.212
16	1.168	1.207
17	1.216	1.210
18	1.178	1.208
19	1.258	1.212
20	1.221	1.210
Mean	1.210	1.210
Median	1.216	1.210
Std Dev.	0.031	0.002
Rel.Std.Dev.	2.52%	0.135%

# Table 5. Neutron Activation Analysis of Au (in ppm) on 20 x 85mg subsamples and showing theequivalent results scaled to a 30g sample mass typical of fire assay determination.

\*Results calculated for a 30g equivalent sample mass using the formula:  $x^{30g Eq} = \frac{(x^{INA} - \bar{x}) \times RSD@30g}{RSD@85mg} + \bar{X}$ where  $x^{30g Eq}$  = equivalent result calculated for a 30g sample mass

 $(x^{INAA}) = raw INAA result at 85mg$ 

 $\bar{X}$  = mean of 85mg INAA results

The homogeneity of OREAS 234 has also been evaluated in a **nested ANOVA** of the round robin program. Each of the forty-two round robin laboratories received six samples per CRM and these samples were made up of paired samples from three different, non-adjacent sampling intervals. The purpose of the ANOVA evaluation is to test that no statistically significant difference exists in the variance between units to that of the variance within units. This allows an assessment of homogeneity across the entire prepared batch of OREAS 234. The test was performed using the following parameters:

- Gold fire assay 186 samples (31 laboratories each providing analyses on 3 pairs of samples);
- Gold aqua regia digestion 132 samples (22 laboratories each providing analyses on 3 pairs of samples);
- Gold cyanide leach 120 samples (20 laboratories each providing analyses on 3 pairs of samples);
- Null Hypothesis, H<sub>0</sub>: Between-unit variance is no greater than within-unit variance (reject H<sub>0</sub> if *p*-value < 0.05);</li>



• Alternative Hypothesis, H<sub>1</sub>: Between-unit variance is greater than within-unit variance.

*P*-values are a measure of probability where values less than 0.05 indicate a greater than 95% probability that the observed differences in within-unit and between-unit variances are real. The datasets were filtered for both individual and laboratory data set (batch) outliers prior to the calculation of the *p*-value.

This process derived *p*-values of 0.99 for Au by fire assay, 0.57 for Au by aqua regia digestion and 0.27 for Au by cyanide leach. All *p*-values are insignificant and the Null Hypothesis is retained. Additionally, none of the other certified values showed significant *p*-values.

Only results for constituents present in concentrations well above the detection levels (i.e. >20 x Lower Limit of Detection) for the various methods undertaken were considered for the objective of evaluating homogeneity. It is important to note that ANOVA is not an absolute measure of homogeneity. Rather, it establishes whether or not the analytes are distributed in a similar manner throughout the packaging run of OREAS 234 and whether the variance between two subsamples from the same unit is statistically distinguishable from the variance of two subsamples taken from any two separate units. A reference material therefore can possess poor absolute homogeneity yet still pass a relative homogeneity (ANOVA) test if the within-unit heterogeneity is large and similar across all units.

Based on the statistical analysis of the results of the inter-laboratory certification program it can be concluded that OREAS 234 is fit-for-purpose as a certified reference material (see 'Intended Use' below).

# PARTICIPATING LABORATORIES

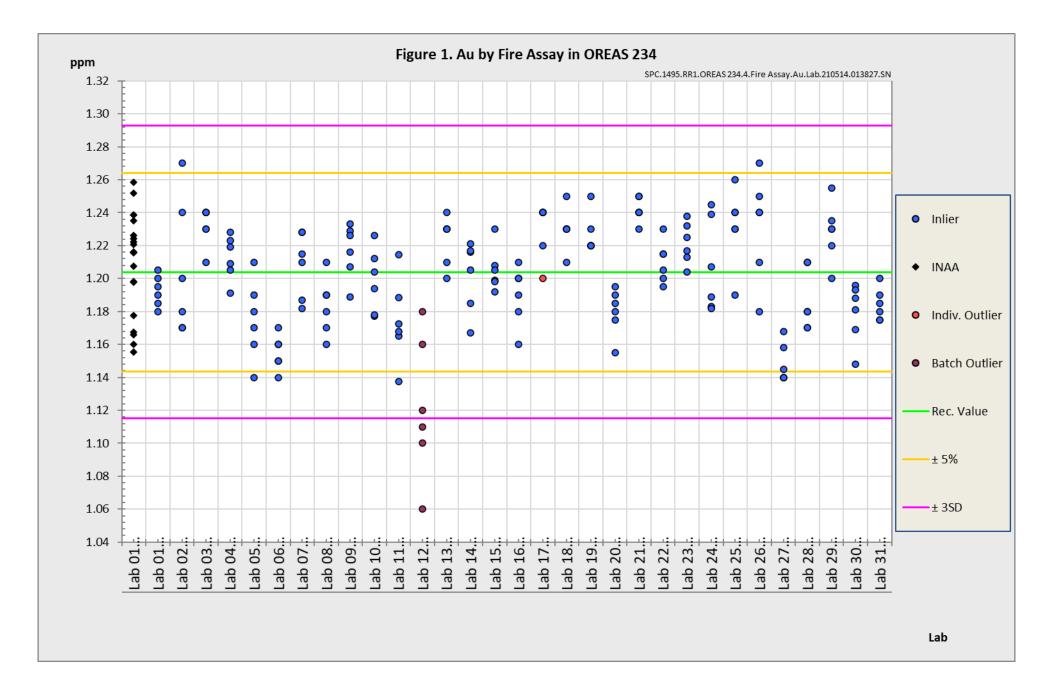
- 1. Actlabs, Ancaster, Ontario, Canada
- 2. AGAT Laboratories, Mississauga, Ontario, Canada
- 3. Alex Stewart International, Mendoza, Argentina
- 4. ALS, Lima, Peru
- 5. ALS, Loughrea, Galway, Ireland
- 6. ALS, Perth, WA, Australia
- 7. ALS, Reno, Nevada, USA
- 8. ALS, Vancouver, BC, Canada
- 9. ANSTO, Lucas Heights, NSW, Australia
- 10. Bureau Veritas Commodities and Trade, Inc., Sparks, Nevada, USA
- 11. Bureau Veritas Commodities Canada Ltd, Vancouver, BC, Canada
- 12. Bureau Veritas Geoanalytical, Adelaide, SA, Australia
- 13. Bureau Veritas Geoanalytical, Perth, WA, Australia
- 14. Chrysos Corporation, Kalgoorlie, WA, Australia
- 15. Chrysos Corporation, Perth, WA, Australia
- 16. ESAN Istanbul, Istanbul, Turkey
- 17. Inspectorate (BV), Lima, Peru
- 18. Intertek Genalysis, Adelaide, SA, Australia



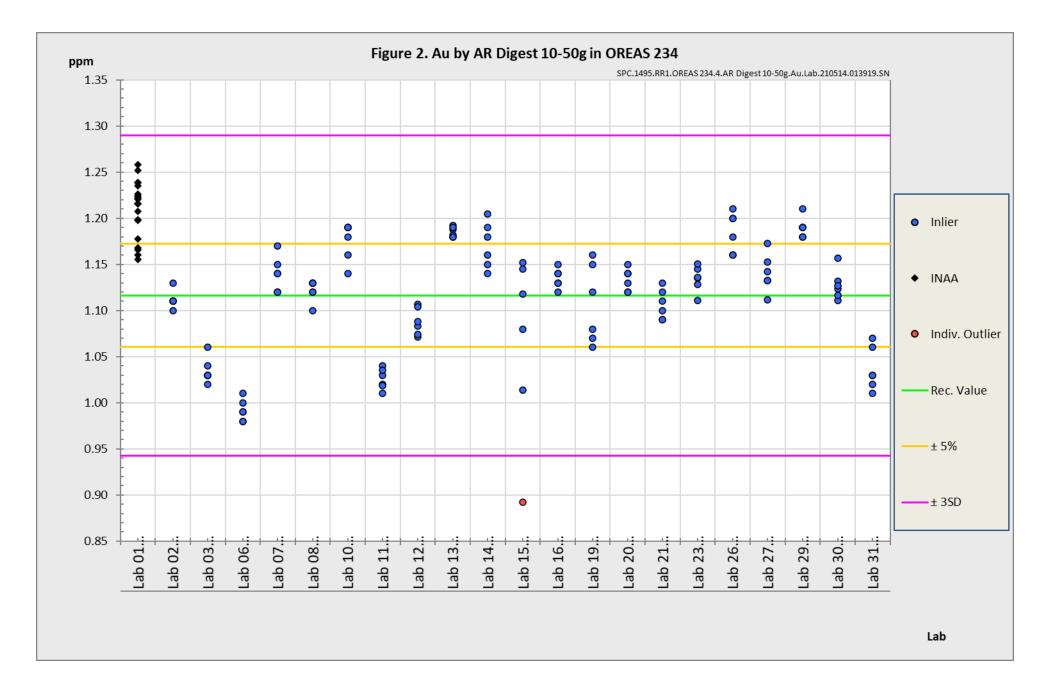
- 19. Intertek Genalysis, Perth, WA, Australia
- 20. Intertek Tarkwa, Tarkwa, Ghana
- 21. Intertek Testing Services Philippines, Cupang, Muntinlupa, Philippines
- 22. MinAnalytical Services, Kalgoorlie, WA, Australia
- 23. MinAnalytical Services, Perth, WA, Australia
- 24. MSALABS, Vancouver, BC, Canada
- 25. Nagrom, Perth, WA, Australia
- 26. On Site Laboratory Services, Bendigo, VIC, Australia
- 27. PT Geoservices Ltd, Cikarang, Jakarta Raya, Indonesia
- 28. PT Intertek Utama Services, Jakarta Timur, DKI Jakarta, Indonesia
- 29. Saskatchewan Research Council, Saskatoon, Saskatchewan, Canada
- 30. SGS, Randfontein, Gauteng, South Africa
- 31. SGS Australia Mineral Services, Kalgoorlie, WA, Australia
- 32. SGS Australia Mineral Services, Perth, WA, Australia
- 33. SGS Canada Inc., Vancouver, BC, Canada
- 34. SGS del Peru, Lima, Peru
- 35. SGS Tarkwa, Tarkwa, Western Region, Ghana
- 36. Skyline Assayers & Laboratories, Tucson, Arizona, USA

Please note: To preserve anonymity, the above numbered alphabetical list of participating laboratories <u>does not</u> correspond with the Lab ID numbering on the scatter plots below.

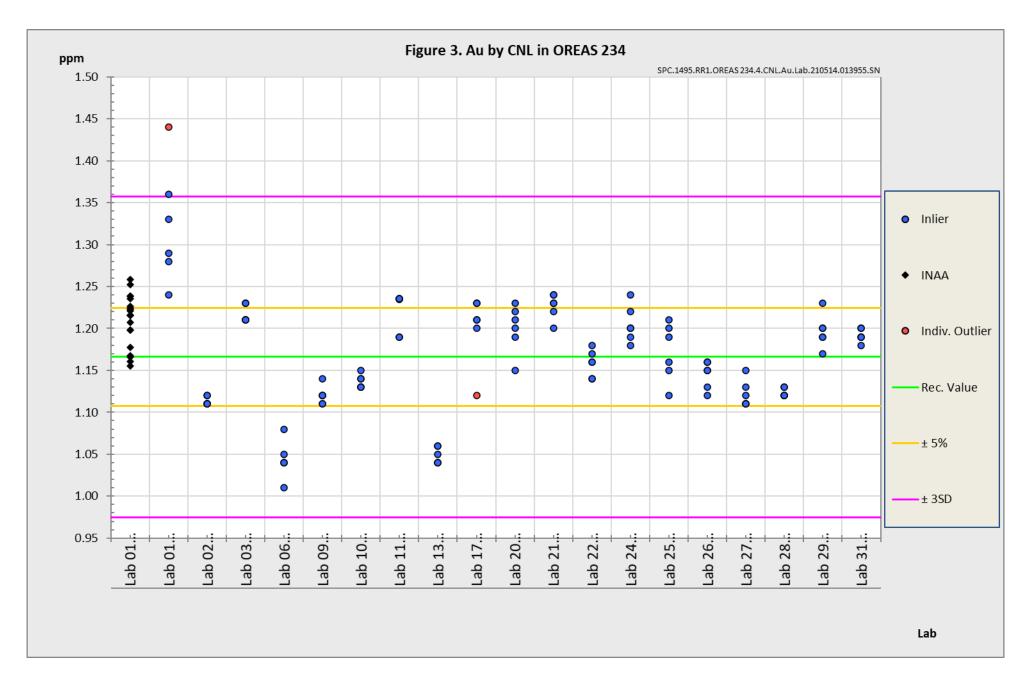














## PREPARER AND SUPPLIER

Certified reference material OREAS 234 was prepared, certified and supplied by:



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# METROLOGICAL TRACEABILITY

The analytical samples were selected in a manner representative of the entire batch of the prepared CRM. This 'representivity' was maintained in each submitted laboratory sample batch and ensures the user that the data is traceable from sample selection through to the analytical results that underlie the consensus values. Each analytical data set has been validated by its assayer through the inclusion of internal reference materials and QC checks during analysis.

The laboratories were chosen on the basis of their competence (from past performance in inter-laboratory programs undertaken by ORE Pty Ltd) for a particular analytical method, analyte or analyte suite and sample matrix. Most of these laboratories have and maintain ISO 17025 accreditation. The certified values presented in this report are calculated from the means of accepted data following robust statistical treatment, as detailed in this report.

Guide ISO/TR 16476:2016, section 5.3.1 describes metrological traceability in reference materials as it pertains to the transformation of the measurand. In this section it states, *"Although the determination of the property value itself can be made traceable to appropriate units through, for example, calibration of the measurement equipment used, steps like the transformation of the sample from one physical (chemical) state to another cannot. Such transformations may only be compared with a reference (when available), or among themselves. For some transformations, reference methods have been defined and may be used in certification projects to evaluate the uncertainty associated with such a transformation. In other cases, only a comparison among different laboratories using the same method is possible. In this case, certification takes place on the basis of agreement among independent measurement results (see ISO Guide 35:2006, Clause 10)."* 

## COMMUTABILITY

The measurements of the results that underlie the certified values contained in this report were undertaken by methods involving pre-treatment (digestion/fusion) of the sample. This served to reduce the sample to a simple and well understood form permitting calibration using simple solutions of the CRM. Due to these methods being well understood and highly effective, commutability is not an issue for this CRM. All OREAS CRMs are sourced from natural ore minerals meaning they will display similar behaviour as routine 'field' samples in the relevant measurement process. Care should be taken to ensure 'matrix matching' as close as practically achievable. The matrix and mineralisation style of the CRM is described in the 'Source Material' section and users should select appropriate CRMs matching these attributes to their field samples.



## INTENDED USE

OREAS 234 is intended to cover all activities needed to produce a measurement result. This includes extraction, possible separation steps and the actual measurement process (the signal producing step). OREAS 234 may be used to calibrate the entire procedure by producing a pure substance CRM transformed into a calibration solution.

OREAS 234 is intended for the following uses:

- For the monitoring of laboratory performance in the analysis of analytes reported in Table 1 in geological samples;
- For the verification of analytical methods for analytes reported in Table 1;
- For the calibration of instruments used in the determination of the concentration of analytes reported in Table 1.

### QC monitoring using multiples of the Standard Deviation (SD)

In the application of SD's in monitoring performance it is important to note that not all laboratories function at the same level of proficiency and that different methods in use at a particular laboratory have differing levels of precision. Each laboratory has its own inherent SD (for a specific concentration level and analyte-method pair) based on the analytical process and this SD is not directly related to the round robin program.

The majority of data generated in the round robin program was produced by a selection of world class laboratories. The SD's thus generated are more constrained than those that would be produced across a randomly selected group of laboratories. To produce more generally achievable SD's the 'pooled' SD's provided in this report include inter-laboratory bias. This 'one size fits all' approach may require revision at the discretion of the QC manager concerned following careful scrutiny of QC control charts.

## STABILITY AND STORAGE INSTRUCTIONS

OREAS 234 is low in reactive sulphide (0.37 wt.% S) and in its unopened state and under normal conditions of storage has a shelf life beyond ten years. Its stability will be monitored at regular intervals and purchasers notified if any changes are observed.

### Single-use sachets

Following analysis of the CRM subsample it is the manufacturers' expectation that any remaining material is discarded. The stability of the material after opening the sachet is not within the scope of proper use. However, if opened sachets are resealed after opening, then under ordinary\* storage conditions the CRM will have a shelf-life beyond ten years.

\*ordinary storage conditions: means storage not in direct sunlight in a dry, clean, well ventilated area at temperatures between -5° and 50°C.

### Repeat-use packaging (e.g., 1kg plastic jars)

The stability of the CRM after opening the lid of the plastic jar is only affected by local atmospheric conditions with regard to oxidation and hygroscopic change. There is no segregation affect (please see our <u>Technical Note on Particle Segregation</u>).



The primary cause of change through oxidation is in relation to the breakdown of sulphide minerals to sulphates and is negligible for OREAS 234 given its low sulphur concentration (0.19 wt.% S).

Hygroscopic change is the amount of absorbed moisture (weakly held H<sub>2</sub>O- molecules on the surface of exposed material) following exposure to the local atmosphere. Usually, equilibration of material to the local atmosphere will only occur if the material is spread into a thin (~2mm thick) layer and left exposed for a period of 2 hours. OREAS 234 contains a non-hygroscopic matrix and therefore, exposure to a local atmosphere that is significantly different (in terms of temperature and humidity) from the climate during manufacturing will have negligible impact on the precision of results. The 'Physical Properties' section indicates the approximate moisture concentration.

# **INSTRUCTIONS FOR CORRECT USE**

The certified values for OREAS 234 refer to the concentration level in its packaged state. It should not be dried prior to weighing and analysis. 1kg jars permit repeated sampling as long as the lid is promptly re-secured to prevent airborne contamination.

### Minimum sample size

As a practical guide, the minimum mass of sample used should match the typical mass that the laboratories used in the interlaboratory (round robin) certification program. This means that different sample masses should be used depending on the operationally defined methodology.

- Au by fire assay: ≥30g;
- Au by aqua regia digestion: ≥25g;
- Au by cyanide leach: ≥20g;
- Au by PhotonAssay: ~350g;
- 4-acid digestion with ICP-OES and/or MS finish: ≥0.25g;
- Aqua regia digestion with ICP-OES and/or MS finish: ≥0.5g.

### **QC** monitoring using multiples of the Standard Deviation (SD)

In the application of SD's in monitoring performance it is important to note that not all laboratories function at the same level of proficiency and that different methods in use at a particular laboratory have differing levels of precision. Each laboratory has its own inherent SD (for a specific concentration level and analyte-method pair) based on the analytical process and this SD is not directly related to the round robin program.

The majority of data generated in the round robin program was produced by a selection of world class laboratories. The SD's thus generated are more constrained than those that would be produced across a randomly selected group of laboratories. To produce more generally achievable SD's the 'pooled' SD's provided in this report include inter-laboratory bias. This 'one size fits all' approach may require revision at the discretion of the QC manager concerned following careful scrutiny of QC control charts.

The performance gates shown in Table 1 are intended only to be used as a first principle guide as to what a laboratory may be able to achieve. Over a period of time monitoring your own laboratory's data for this CRM, SDs should be calculated directly from your own laboratory's process. This will enable you to establish more specific performance gates



that are fit for purpose for your application as well as the ability to monitor bias. If your long-term trend analysis shows an average value that is within the 95% confidence interval then generally there is no cause for concern in regard to bias.

It is important to note that in the analytical industry there is no standardisation of the aqua regia digestion process. This method is a partial empirical digest and differences in recoveries for various analytes are commonplace. These are caused by variations in the digest conditions and can include the ratio of nitric to hydrochloric acids, acid strength, temperatures, leach times and secondary digestions. Recoveries for sulphide-hosted base metal sulphides approach total values, however, other analytes, in particular the lithophile elements, show greater sensitivity to method parameters. This can result in lack of consensus in an inter-laboratory certification program for these elements.

The approach applied here is to report certified values in those instances where reasonable agreement exists amongst a majority of participating laboratories. The results of specific laboratories may differ significantly from the certified values, but will, nonetheless, be valid and reproducible in the context of the specifics of the aqua regia method in use. Users of this reference material should, therefore, be mindful of this limitation when applying the certified values in a quality control program.

### HANDLING INSTRUCTIONS

Fine powders pose a risk to eyes and lungs and therefore standard precautions including the use of safety glasses and dust masks are advised.

# LEGAL NOTICE

Ore Research & Exploration Pty Ltd has prepared and statistically evaluated the property values of this reference material to the best of its ability. The Purchaser by receipt hereof releases and indemnifies Ore Research & Exploration Pty Ltd from and against all liability and costs arising from the use of this material and information.

## **DOCUMENT HISTORY**

Revision No.	Date	Changes applied
0	14 <sup>th</sup> May, 2020	First publication.

# **CERTIFYING OFFICER**

14<sup>th</sup> May, 2021

Craig Hamlyn (B.Sc. Hons - Geology), Technical Manager - ORE P/L



## **QMS CERTIFICATION**

ORE Pty Ltd is ISO 9001:2015 certified by Lloyd's Register Quality Assurance Ltd for its quality management system including development, manufacturing, certification and supply of CRMs.



# REFERENCES

Govett, G.J.S. (1983). Handbook of Exploration Geochemistry, Volume 2: Statistics and Data Analysis in Geochemical Prospecting (Variations of accuracy and precision).

Ingamells, C. O. and Switzer, P. (1973). Talanta 20, 547-568.

ISO Guide 30:2015. Terms and definitions used in connection with reference materials.

ISO Guide 31:2015. Reference materials – Contents of certificates and labels.

ISO Guide 35:2017. Certification of reference materials - General and statistical principals.

ISO 16269:2014. Statistical interpretation of data – Part 6: Determination of statistical tolerance intervals.

ISO/TR 16476:2016, Reference Materials – Establishing and expressing metrological traceability of quantity values assigned to reference materials.

ISO 17025:2005, General requirements for the competence of testing and calibration laboratories.

Munsell Rock Color Book (2014). Rock-Color Chart Committee, Geological Society of America (GSA), 4300 44th Street SE, Grand Rapids, MI 49512.

